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# On the structural and magnetic properties of the ternary silicides $Ce_6M_{1,67}Si_3$ (M = Co, Ni) and $Ce_5Ni_{1,85}Si_3$

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#### Abstract

Contrary to that reported previously, the ternary silicide " $Ce_6Ni_2Si_3$ " does not exist. The melting of this alloy, followed or not by annealing, leads to the existence of the two new ternary compounds,  $Ce_5Ni_{1.67}Si_3$  and  $Ce_5Ni_{1.85}Si_3$ . The investigation of these ternary silicides based on nickel and Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> by X-ray diffraction on single crystal reveals an ordered distribution between Ni (or Co) and Si atoms. The nickel or cobalt positions in the chains of face-shared octahedra of cerium are not fully occupied with a strong delocalisation of their electron density. The structural investigations of these compounds confirm that the "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" and "Ce<sub>5</sub>Ni<sub>2</sub>Si<sub>3</sub>" structural type have to be rewritten as  $Ce_6Ni_{2-x}Si_3$  and  $Ce_5Ni_{2-x}Si_3$ . Magnetisation and specific heat measurements evidence a magnetic ordering at 3.8(2) K for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> and a heavy fermion behaviour for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>. © 2007 Elsevier Inc. All rights reserved.

Keywords: Crystal structure; Disordered structures; Magnetic order; Heavy fermion

## 1. Introduction

The family of ternary silicides  $RE_6Ni_2Si_3$  with RE = La, Ce, Pr and Nd was discovered by Bodak et al. [1]. Using X-ray diffraction films, these authors have determined the structure of the so-called "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" compound which crystallises with the space group  $P6_3/m$ . Its structure is characterised by infinite chains of face-shared [Ce<sub>6</sub>] octahedra running along the c-axis and triangular columns of centred [Ce<sub>6</sub>] trigonal prisms sharing faces. Ni atoms occupy the centres of the octahedra, whereas the centre of trigonal prisms is occupied in part by Si atoms alone and in part by an Ni/Si (2Ni/4Si) mixture. Moreover, this structure leads to very short Ni-Ni distances (2.162 Å compared to the metallic radius  $r_{\rm Ni} = 1.246 \,\text{\AA}$ ) between two  $[Ce_6]$  octahedra along the *c*-axis.

Recently, the ternary silicides La<sub>6</sub>Ni<sub>1.54</sub>Si<sub>3</sub> and Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub> deriving from the Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>-type structure

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were prepared [2,3]. In these structural determinations, it was reported that Ni-Si sublattice forms an ordered distribution. Also, it was claimed that the nickel sites on the  $6_3$  axes (in the [La<sub>6</sub> or Nd<sub>6</sub>] octahedra) are not fully occupied and shifted in two positions for steric reasons. This structural model leads to Ni-Ni distances (2.818Å) between two [Nd<sub>6</sub>] octahedra in Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub>, compatible with the metallic radius of nickel [3].

More recently, we have reported on the existence of the ternary silicide based on cobalt Gd<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> that orders ferromagnetically at 294 K [4] and is isotypic to La<sub>6</sub>Ni<sub>1.54</sub>Si<sub>3</sub> and Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub>. Moreover, extended study leads to synthesise the new ternary compounds  $Nd_6Co_{1.67}$  $Si_3$  and  $Tb_6Co_{1.67}Si_3$  [5]. All these results suggest that the composition "Ce6Ni2Si3" could not exist as single phase. In order to answer this question, we have prepared and characterised by X-ray diffraction on single crystal, the alloys "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" and Ce<sub>6</sub> $M_{1.67}$ Si<sub>3</sub> (M =Co, Ni). We present here the crystallographic and magnetic properties of Ce<sub>6</sub>M<sub>1.67</sub>Si<sub>3</sub> and Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub>, which are obtained by annealing of "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" composition at 1073 K.

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## 2. Experimental details

## 2.1. Synthesis

The polycrystalline  $Ce_6M_{1.67}Si_3$  (M = Co, Ni) and "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" samples were synthesised by arc melting a stoichiometric mixture of pure elements (purity above 99.9%) in a high-purity argon atmosphere. Then, the samples were turned over and remelted several times to ensure homogeneity. The weight loss during the arcmelting process was less than 0.5 wt%. Annealing was done for 1 month at 1073 K by enclosing the samples in evacuated quartz tubes. No attack of the quartz tubes by the samples and vice versa was observed. The stoichiometry deduced from the X-ray diffraction study of the single crystal of the phase Ce<sub>5</sub>Ni<sub>2-x</sub>Si<sub>3</sub> found in the annealed alloy of "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" is Ce<sub>5</sub>Ni<sub>185</sub>Si<sub>3</sub> (see below). A polycrystalline sample of this last ternary silicide has been prepared for magnetic measurements with the same procedure as the one used for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>.

The as-cast and annealed samples of "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" and Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> were checked by microprobe analysis using a Cameca SX-100 instrument. The analysis was performed on the basis of intensity measurements of CeL $\alpha_I$ , Ni $K\alpha_I$  and Si $K\alpha_I$  X-ray emission lines, which were compared with those obtained for the equiatomic ternary silicide CeNiSi used as reference compound.

X-ray powder diffraction with the use of a Philips 1050diffractometer (CuK $\alpha$  radiation) was applied, before and after annealing, for the characterisation of the structural type and the phase identification of the samples.

#### 2.2. X-ray diffraction on single crystal

Suitable single crystals of  $Ce_6Ni_{1.67}Si_3$ ,  $Ce_6Co_{1.67}Si_3$  and  $Ce_5Ni_{1.85}Si_3$  were selected by optical microscopy. Reflection data were collected on an Enraf-Nonius Kappa CCD area-detector diffractometer at room temperature using  $MoK\alpha$  radiation. The single crystals of  $Ce_5Ni_{1.85}Si_3$  were found in the annealed alloy of " $Ce_6Ni_2Si_3$ ". Many crystals of  $Ce_6Co_{1.67}Si_3$  have been tested because of their poor quality, the best one being kept for the final data collection. A Gaussian-type absorption correction was applied, the shape of the crystals being determined with the video microscope of the diffractometer. Data processing and all refinements were performed with the JANA2000 program package [6]. Details of data collections and structure refinements are listed in Table 1.

## 2.3. Magnetisation and specific heat measurements

Magnetisation measurements were performed using a Superconducting QUantum Interference Device magnetometer in the temperature range 1.8–370 K and applied fields up to 5 T. Heat capacity measurements were realised by a relaxation method with a Quantum Design PPMS system and using a two tau model analysis.

## 3. Results and discussion

# 3.1. Existence of phases

The analysis of "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" as-cast sample by X-ray powder diffraction reveals the presence of two phases (Fig. 1): (i) one main phase which crystallises in the hexagonal unit with cell parameters a = 12.06(2) and c = 4.31(1)Å and (ii) the binary compound Ce<sub>7</sub>Ni<sub>3</sub> [7]. This result is confirmed by microprobe analysis of the sample. These phases are clearly distinguished on the characteristic microstructure (back-scattered image) presented in Fig. 2: (i) one showing white grey ranges, which corresponds to Ce<sub>7</sub>Ni<sub>3</sub> and (ii) the second (grey) having Ce 56.1(4) %, Ni 15.1(3) % and Si 28.8(4) % as experimental atomic percentages. This last chemical composition is typical to the Ce<sub>6</sub>Ni<sub>1 61(3)</sub>Si<sub>3 08(4)</sub> ternary silicide. In other words, the study indicates that "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" (Ce 54.5%, Ni 18.2% and Si 27.3%) cannot be obtained as single phase by melting followed by quenching.

Moreover, the X-ray powder pattern of "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" annealed sample (Fig. 1) shows the occurrence of another phase; the main peaks attributed to this phase can be indexed with the hexagonal unit cell parameters a = 16.01(2) and c = 4.26(1) Å. The microprobe analysis of this last phase (Ce 50.8(4)%, Ni 18.2(3)% and Si 31.0(4)%) suggests the presence of Ce<sub>5</sub>Ni<sub>1.79(3)</sub>Si<sub>3.05(4)</sub> in the annealed sample. It should be noted that two isotype compounds La<sub>5</sub>Ni<sub>1.75</sub>Si<sub>3</sub> and Pr<sub>5</sub>Ni<sub>1.9</sub>Si<sub>3</sub> were prepared recently [2,8]. In conclusion, the ternary silicide "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" does not exist but the characterisation of this alloy (obtained after melting and quenching or after annealing at 1073 K) indicates the existence of two ternary compounds Ce<sub>6</sub>Ni<sub>1.61(3)</sub>Si<sub>3.08(4)</sub> and Ce<sub>5</sub>Ni<sub>1.79(3)</sub>Si<sub>3.05(4)</sub>.

On the contrary, X-ray powder diffraction analysis performed on as-cast or annealed Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> sample (Fig. 3) reveals no phase coexistence; these patterns are indexed with the hexagonal unit cell parameters a = 12.06(2) and c = 4.31(1)Å. After melting, some traces of Ce<sub>5</sub>Si<sub>3</sub> binary silicide are detected [9] but disappear after annealing. These results show that Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>, detected by X-ray powder diffraction and microprobe analysis in the "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" as-cast alloy, is a new member of the family  $RE_6Ni_{1.67-x}Si_3$  reported previously for RE = Laand Nd [2,3]. The synthesis of the hexagonal phase Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> with the stoichiometry deduced from the single-crystal XRD study (see below) led to a pure sample with cell parameters a = 16.011(6) and c = 4.263(5) Å. This stoichiometry is very close to the one deduced from the microprobe analysis.

Similar investigation performed on Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> sample reveals the existence of this new ternary silicide crystallising as Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> in the hexagonal unit cell (a = 12.04(2) and c = 4.27(1)Å). The compounds  $RE_6$ Co<sub>1.67</sub>Si<sub>3</sub> prepared recently by us for RE =Gd [4], can be obtained for instance with RE = Nd and Tb [5].

Table 1 Crystal data and structure refinement for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>, Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> and Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>

Composition	$Ce_6Ni_{1.68(8)}Si_3$	$Ce_5Ni_{1.851(9)}Si_3$	Ce <sub>6</sub> Co <sub>1.63(10)</sub> Si <sub>3</sub>
Crystal data			
Chemical formula	Ce <sub>6</sub> Ni <sub>1.67</sub> Si <sub>3</sub>	Ce <sub>5</sub> Ni <sub>1.85</sub> Si <sub>3</sub>	$Ce_6Co_{1.67}Si_3$
M <sub>r</sub>	1023	892.4	1023.4
Cell setting, space group		Hexagonal, $P6_3/m$	
u,c (Å)	12.0506(10), 4.3030(2)	16.0014(15), 4.2701(2)	12.039(2), 4.2624(6)
$V(Å^3)$	541.15(7)	946.86 (13)	535.01(15)
Z	2	4	2
$D_{x} (Mg m^{-3})$	6.276	6.258	6.351
Radiation type		Μο <i>Κ</i> α	
No. of reflections for cell parameters	1522	393	393
range (deg)		5-40	
$\iota (\mathrm{mm}^{-1})$	27.78	27.38	27.75
Cemperature (K)		293 K	
Crystal form, colour		Block, metallic light grey	
Crystal size (mm)	$0.008 \times 0.018 \times 0.05$	$0.08 \times 0.04 \times 0.018$	$0.035 \times 0.018 \times 0.014$
Data collection Diffractometer		Nonius KappaCCD	
Data collection method		$\phi$ and $\omega$ frames	
Absorption correction		Gaussian	
*	0.201	0.238	0.453
T <sub>min</sub>	0.647	0.238	0.433
$T_{\rm max}$ No. of measured, independent and observed	10285, 1220, 974	19367, 2139, 1675	7087, 1211, 536
eflections	10283, 1220, 974	19507, 2159, 1075	/08/, 1211, 550
Criterion for observed reflections	$I > 2\sigma(I)$	$I > 2\sigma(I)$	$I > 3\sigma(I)$
	0.071, 0.076	1 > 20(1) 0.079, 0.085	0.097, 0.183
$R_{int}$ (obs), $R_{int}$ (all)	0.071, 0.076	40.0	0.097, 0.185
max (°) Range of <i>hkl</i>	$-21 \rightarrow h \rightarrow 19$	$-28 \rightarrow h \rightarrow 28$	$-21 \rightarrow h \rightarrow 20$
cange of <i>n k i</i>	$-21 \rightarrow h \rightarrow 19$ $-21 \rightarrow k \rightarrow 21$	$-28 \rightarrow h \rightarrow 28$ $-28 \rightarrow k \rightarrow 28$	$-21 \rightarrow h \rightarrow 20$ $-21 \rightarrow k \rightarrow 21$
	$-21 \rightarrow k \rightarrow 21$ $-6 \rightarrow l \rightarrow 7$	$-2\delta \rightarrow k \rightarrow 2\delta$ $-7 \rightarrow l \rightarrow 7$	$-21 \rightarrow k \rightarrow 21$ $-7 \rightarrow l \rightarrow 3$
	$-0 \rightarrow l \rightarrow l$	$-/ \rightarrow l \rightarrow l$	$-1 \rightarrow l \rightarrow 3$
Refinement		_	
Refinement on		$F^2$	
$R[F^2 > 2\sigma(F^2)], WR(F^2), S$	0.039, 0.076, 1.20	0.035, 0.072, 1.11	0.049, 0.114, 1.12
No. of reflections	1220 reflections	2139 reflections	1211 reflections
No. of parameters	31	44	30
Weighting scheme		Based on measured s.u.'s	
		$w = 1/(\sigma^2(I) + 0.0009I^2)$	
$\Delta/\sigma)_{ m max}$		0.001	
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} \ ({\rm e}  {\rm \AA}^{-3})$	2.81, -2.49	3.26, -2.78	3.75, -3.02
Extinction method		Type 1 Lorentzian isotropic	None
		(Becker & Coppens, 1974)	
Extinction coefficient	0.082 (16)	0.31(3)	

Computer programs: (see ref. [6]).

# 3.2. Crystal structure

# 3.2.1. Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>

The space group  $P6_3/m$ , previously used for the refinement of Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub> and the homologous compounds La<sub>6</sub>Ni<sub>1.54</sub>Si<sub>3</sub> and Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub> [1–3], was found to be compatible with the observed extinction rules for Ce<sub>6</sub>Ni<sub>1.67</sub> Si<sub>3</sub>. The starting atomic coordinates were determined with direct method. The Ce1 and Ce2 positions correspond to the *6h* Wyckoff positions (Table 2), silicon and nickel (Ni2) atoms are ordered over the *6h* and *2d* sites. When the Ce1, Ce2, Ni2 and Si atoms were introduced, the difference-Fourier maps analysis showed clearly significant residues in (001/4). Then, the Ni3 atom has been placed at this last position and its occupancy factor refined but small residues

in the difference-Fourier maps were found around the site (000). The Ni1 position was placed with coordinates (000) and the occupancy factor refined. When anisotropic atomic displacement parameters (ADP) were introduced, non-definite positive ADP matrix was observed for Ni1 position with very high value of  $U^{33}$ . To avoid this problem, the Ni1 position has been shifted away from (000) (Table 2). The use of anisotropic ADP for all positions and refined occupancy factors of Ni1 and Ni3 positions led to reliability factors of  $R/R_w(all) = 0.039/0.076$ . This structural model is very close to the ones proposed by Prots and Jeitschko [2,3] for La<sub>6</sub>Ni<sub>1.54</sub>Si<sub>3</sub> and Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub>. Occupancy factors of all positions have been released to be sure that the distribution between Ni and Si is correct and then all occupancy factors have been fixed to their ideal value,

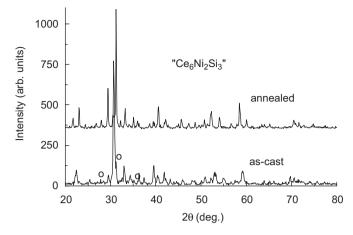


Fig. 1. X-ray powder patterns of " $Ce_6Ni_2Si_3$ " alloy obtained by melting (as-cast) or annealing (o indicates the peak position for  $Ce_7Ni_3$ ).

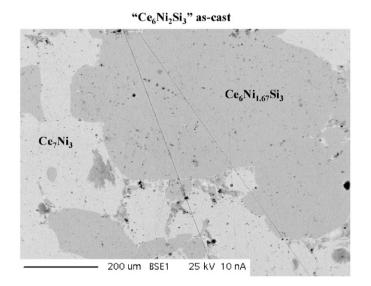


Fig. 2. Microstructure of "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" alloy obtained after melting and quenching. The two phases Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> (grey) and Ce<sub>7</sub>Ni<sub>3</sub> (white grey) are indicated.

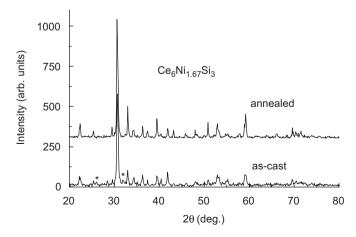


Fig. 3. X-ray powder patterns of  $Ce_6Ni_{1.67}Si_3$  ternary silicide obtained by melting (as-cast) or annealing (\* indicates the peak position for  $Ce_5Si_3$ ).

except for Ni1 and Ni3 atoms. As previously pointed by Prots and Jeitschko [2,3], the description of the delocalisation of the electron density along the chain in (00z) is a critical point in this structural type. To better understand this problem, the next step of our refinement needs to be detailed.

The Fourier map (Fig. 4) calculated after the first refinement (vide supra) shows the strong delocalisation of the electron density along the (00z) direction. Moreover, a significant delocalisation of the Ni3 density in the *ab* plane (Fig. 4) is observed. This delocalisation is not strong enough to be described with splitting positions then a development, up to third order, of ADP for Ni3 position was introduced. This last model leads to lower difference-Fourier residues. Two contour maps of the joint probability density function (j.p.d.f.) of the Ni1 and Ni3 positions have been drawn in Fig. 5. These maps are in good agreement with the Fourier maps (Fig. 4) except the fact that no maximum corresponding to the Nil position is observed in the Fourier map. The introduction of the Nil position is necessary to describe the delocalisation of the electron density along the (00z)-axis. The main position for nickel atoms in the chain is Ni3 (Table 2). The delocalisation of the Ni3 position in the ab plane induces shorter distances with the neighbourhood Cel atoms. This steric strain explains the significant deformation of the thermal ellipsoid of the Cel positions in the *ab* plane (Table 3). The motion or vibration of cerium atoms in the *ab* plane is possible because of empty space in the structure (Fig. 6).

The formula deduced from the single-crystal data refinement, Ce<sub>6</sub>Ni<sub>1.68(8)</sub>Si<sub>3</sub> (Table 2), is very close to the one deduced from the microprobe analysis, Ce<sub>6</sub>Ni<sub>1.61(3)</sub>  $Si_{3,08(4)}$  This composition is significantly different from the ideal one, Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>, proposed by Bodak et al. [1]. This latter composition is not compatible with the steric strains in this structural type. Indeed, as firstly proposed by Lemaire et al. [10] for  $RE_4Co_{3.07}$  (or  $RE_6Co_{4.61}$ ) series (RE = rare earth), the steric strains between positions of Co in the chain (00z) imply a non-stoichiometry of these phases, empty space are needed to manage realistic Co-Co distances. The overall formula of the series  $RE_6Co_{4.61}$  can be expressed as  $RE_6Co_{1.61}Co_3$ , to emphasise the structural similarity with Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>. These authors proposed the limit composition  $RE_6Co_{1.67}Co_3$  for this series and they have observed such limit when they have attempted to prepare Ho<sub>6</sub>Co<sub>1.75</sub>Co<sub>3</sub>. In their model, they have observed only one site for Co in the chain (00z), this site corresponding to the Nil site of our model (same notation for Ref. [2,3]). If two third of this position are occupied by Co atoms, a shift away from the ideal position (000) allows realistic distances between Co atoms. In Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>, the same discussion about an ideal distribution of Ni atoms can be done by considering only the Ni3 position. This position is strongly delocalised along the z-axis to manage realistic Ni-Ni distances. One can observe that our chemical composition deduced from microprobe analysis and structural refinement and the compositions reported

	Atom	Site	x	У	Ζ	$B_{ m eq}$	Occ.
Ce <sub>6</sub> Ni <sub>1.68(8)</sub> Si <sub>3</sub>	Cel	6 <i>h</i>	0.23963(4)	0.22978(4)	1/4	1.527(12)	1
	Ce2	6 <i>h</i>	0.51905(3)	0.13869(3)	1/4	0.936(9)	1
	Ni1	4 <i>e</i>	0	0	0.104(15)	3.3(9)	0.13(2)
	Ni2	2d	1/3	2/3	1/4	1.08(2)	1
	Ni3	2a	0	0	1/4	4.7(5)	0.42(4)
	Si	6 <i>h</i>	0.1610(2)	0.4437(2)	1/4	1.08(5)	1
Ce <sub>6</sub> Co <sub>1.63(10)</sub> Si <sub>3</sub>	Cel	6 <i>h</i>	0.23737(9)	0.22917(10)	1/4	2.01(3)	1
	Ce2	6 <i>h</i>	0.51972(7)	0.14029(7)	1/4	1.02(2)	1
	Col	4 <i>e</i>	0	0	0.07(2)	2.8(13)	0.07(2)
	Co2	2c	1/3	2/3	1/4	1.44(6)	1
	Co3	2a	0	0	1/4	2.1(10)	0.21(4)
	Co4	6h	0.061(2)	0.061(2)	1/4	1.01(2)	0.092(6)
	Si	6 <i>h</i>	0.1599(4)	0.4427(3)	1/4	1.18(10)	1
Ce <sub>5</sub> Ni <sub>1.851(9)</sub> Si <sub>3</sub>	Cel	6 <i>h</i>	0.01060(2)	0.18140(3)	1/4	1.413(9)	1
	Ce2	6 <i>h</i>	0.39891(2)	0.26245(2)	1/4	0.808(8)	1
	Ce3	6h	0.45462(2)	0.06706(2)	1/4	0.794(8)	1
	Ce4	2d	2/3	1/3	1/4	0.775(8)	1
	Ni1	6h	0.28238(5)	0.49837(5)	1/4	0.87(2)	1
	Ni2	2b	0	0	0	5.0(2)	0.193(8)
	Ni3	2a	0	0	1/4	5.0	0.509(10)
	Sil	6h	0.16578(11)	0.55041(11)	1/4	0.84(4)	1
	Si2	6h	0.23288(12)	0.32685(11)	1/4	0.92(4)	1

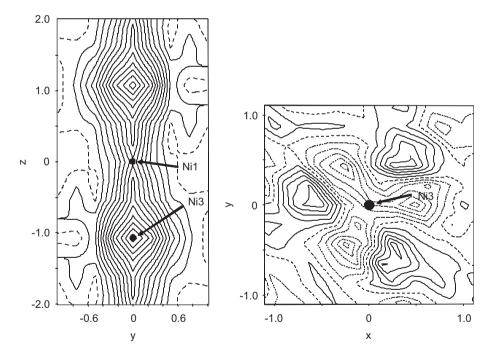


Fig. 4. Structural model of  $Ce_6Ni_{1.67}Si_3$  with anisotropic ADP for all positions: (left) yz section at x = 0 of Fourier map and (right) xy section at z = 1/4 of difference-Fourier maps (contours with full line for positive values and dashed line for negative values in intervals of  $2e \text{ Å}^{-3}$  for Fourier and  $0.4e \text{ Å}^{-3}$  for difference-Fourier map. The filled circles correspond to the refined Ni positions.

from structural refinements by Prots and Jeitschko [2,3] are in good agreement with the limit composition proposed by Lemaire et al. [10]. Prots and Jeitschko have proposed a relationship between the occupancy ratio of Ni1 and Ni3 and the overall composition. They assumed that the maximum number of Ni in the chain (00z) is directly connected to the ratio between occupancy of Ni1 and Ni3 positions [2]. This would imply that for all ternary compounds with composition  $RE_6Ni_{1.67}Si_3$ , the same ratio should be observed. This is not the case in the sequence

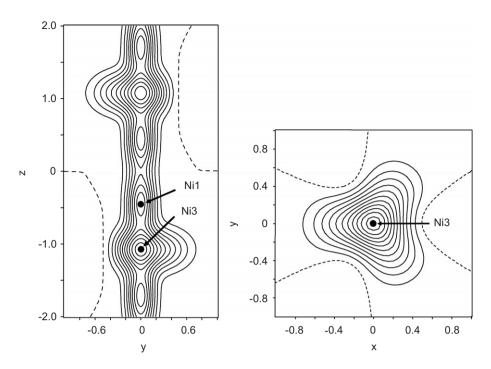


Fig. 5. Joint probability density function maps of Ni for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>: (left) non-harmonic model, *yz* section at x = 0 and (right) non-harmonic model, *xy* section at z = 1/4 (positive (full lines) and negative (dashed lines) contours are drawn in intervals of 0.2 and 0.2 Å<sup>-3</sup>). The filled circles correspond to the refined Ni positions.

Table 3 Anisotropic displacement parameters of Ce1 position in the three ternary silicides

	Atom	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Ce <sub>6</sub> Ni <sub>1.67</sub> Si <sub>3</sub>	Cel	0.0224(2)	0.0291(2)	0.0143(2)	0.0187(2)	0	0
Ce <sub>6</sub> Co <sub>1.67</sub> Si <sub>3</sub>	Cel	0.0329(5)	0.0440(6)	0.0144(3)	0.0304(5)	0	0
$Ce_5Ni_{1.85}Si_3$	Cel	0.01286(14)	0.0282(2)	0.01411(14)	0.01138(13)	0	0

 $La_6Ni_{1.54}Si_3 \rightarrow Ce_6Ni_{1.67}Si_3 \rightarrow Nd_6Ni_{1.66}Si_3$ . The 4e site (00z) of the Nil position is more and more occupied; its occupancy ratio is 9.4%, 13% and 20.3% respectively for ternary compounds based on La, Ce and Nd. On the contrary, the occupancy of the 2a site (001/4), where Ni3 atom is located, decreases in this sequence (42% in Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> and 25% in Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub>; it is difficult to consider this occupancy (35.5%) for  $La_6Ni_{1.54}Si_3$  which exhibits a smaller composition in Ni). The competition between the filling of the 4e and 2a sites by Ni atom can be explained on the basis of steric considerations. The interatomic distances RE1-Ni3 (2.830 Å in Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>) are really short and smaller than the RE1-Ni1 distance  $(2.903 \text{ \AA in Ce}_6\text{Ni}_{1.67}\text{Si}_3)$  (Table 4); when the steric strain increases with decreasing metallic radius of the rare earth  $(La \rightarrow Ce \rightarrow Nd)$ , the 4e site (Ni1 atom) which favours the greater RE1-Ni1 distances is more and more occupied. In this view and in order to keep the stoichiometry of the ternary silicide, the occupancy of the 2a site (Ni3 atom) decreases. In other words, the Ni distribution in the crystal structure of the  $RE_6Ni_{\simeq 1.67}Si_3$  ternary silicides is connected to the nature of the rare-earth element. This explanation is in complete agreement with the structure of  $Ho_6Co_{1.61}Co_3$ . Indeed, in this binary compound the rare-earth element is small and all the Co atoms located in the chain (00z) are found in the Ni1 position.

The unit cell parameters reported here for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> (Table 1) are smaller than those given for the hypothetic ternary silicide "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>" (a = 12.112 and c = 4.323 Å) [1]. Moreover, these two structures differ also from the Ni–Si sublattice; for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> a perfect distribution between Ni and Si is evidenced (Table 2) whereas for instance the site 6h is statistically occupied by Ni and Si in "Ce<sub>6</sub>Ni<sub>2</sub>Si<sub>3</sub>". On the contrary, the crystal structure of Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> can be considered as isotypic with those found for La<sub>6</sub>Ni<sub>1.54</sub>Si<sub>3</sub> and Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub> [2,3]. The unit cell parameters of Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> and Nd<sub>6</sub>Ni<sub>1.66</sub>Si<sub>3</sub>. This last observation suggests that cerium presents a trivalent state in the ternary silicide.

A projection of the structure along the *c*-axis is displayed in Fig. 6. This structure consists of two different types of chains of face-sharing trigonal prisms of Ce atoms filled by Si or Ni atoms. The NiCe<sub>3</sub> chains filled by Ni2

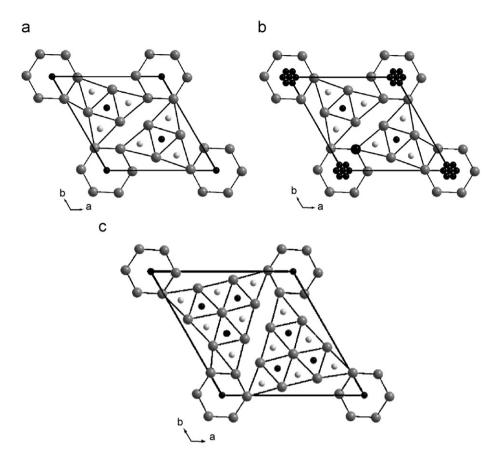


Fig. 6. Projections along the *c*-axis of the structure of (a)  $Ce_6Ni_{1.67}Si_3$ , (b)  $Ce_6Co_{1.67}Si_3$  and (c)  $Ce_5Ni_{1.85}Si_3$ . Ce and Si atoms are represented by large and small grey spheres, respectively, and Ni or Co atoms by black spheres.

atoms are surrounding by three  $SiCe_3$  chains, these chains being connected together by sharing faces. These four trigonal prism chains form triangles in the *ab* plane (Fig. 6). Between six triangles a chain of face-sharing octahedra of Ce is partially filled by Ni atoms (Ni1 and Ni3 positions).

Ni2 atoms are located in trigonal prisms [Ce<sub>6</sub>] with Ni2-Ce distances of 3.020 Å. The distance between Ni2 atoms and neighbourhood Si atoms is equal to 2.440 Å. The Ni2-Ce distances are in good agreement with the sum of metallic radii ( $r_{Ni} = 1.246$ ,  $r_{Ce} = 1.825$  and  $r_{\rm Si} = 1.319$  Å), whereas the Ni2–Si distance is a little bit short. The average Si-Ce distance in the SiCe<sub>3</sub> chain of trigonal prism is equal to 3.126 Å. The trigonal prism of Ce filled by Si is bi-capped by two Ce atoms with Si-Ce distances of 3.160 and 3.297 Å. The Ni1 position is shifted away from the centre of the octahedra of Ce with a long and a short Ni1-Ce distance, 3.21 and 2.903 Å, respectively. The Ni3 position is located in the shared faces of the column of Ce octahedra. This induces very short distance, 2.830 Å, between Ni3 and Ce1 atoms in the *ab* plane. Ce atoms in the two opposite faces of the octahedra form an elongated trigonal prism with long Ni3-Ce1 distances of 3.555 Å. The coordination polyhedra of Ni3 can be described as a tri-capped elongated trigonal prism of Ce atoms.

The coordination polyedra of Ce1 and Ce2 atoms are quite complicated. Only Ce1 position is involved by the disorder in the chain of Ni (00z). Ce1 is surrounded by 3 Si, 10 Ce atoms and by 2 Ni1 and 2 Ni3 positions (Table 4). These latter positions are not fully occupied, then a small statistical disorder around Ce1 position is expected. As previously discussed, very short distance between Ce1 and Ni3 is observed. Ce2 is surrounded by 2 Ni, 5 Si and 10 Ce atoms. All distances between Ce2 and neighbourhood atoms are in good agreement with the sum of metallic radii.

# 3.2.2. Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>

The atomic coordinates determined here for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> have been used as starting point for the refinement of the crystal structure of Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> (space group  $P6_3/m$ ), Ni atoms being replaced by Co atoms. The ordering between Si and Co atoms has been confirmed through occupancy factor refinement. When Co1 and Co3 positions have been introduced to describe the electron density in the chain located at (00z), significant residues in the difference-Fourier map have been observed. The main residues were located in the *ab* plane with z = 1/4 around Co3 position (Fig. 7 (left)). An extra cobalt position (Co4) has been introduced to explain these maxima in the difference-Fourier analysis (Table 2). The strong delocalisation observed in Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> is so strong in Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> that

Table 4 Selected interatomic distances (Å) in Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>, Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> and Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>

			^ 				
Ce1	Ni3 (×1) 2.8302(3)		Si (×2) 3.125(3)		[Ni3 (×1) 1.51(4)]		Ni1 (×2) 2.80(4)
	Ni1 (×2) 2.903(10)		Si ( × 2) 3.169(3)		$[Ni1 (\times 2) 2.15(6)]$		Ce1 ( $\times$ 3) 2.8302(3)
	Si (×2) 3.085(2)		Si (×1) 3.297(2)		Ni3 ( $\times$ 1) 2.80(4)		Ce1 ( $\times$ 6) 3.5551(3)
	Si $(\times 1)$ 3.160(3)		$Ce2 (\times 2) 3.6716(5)$		Ce1 ( $\times$ 3) 2.903(10)		
	Ni1 ( $\times$ 2) 3.21(2)		Ce1 ( $\times$ 2) 3.7074(4)		Ni1 ( $\times$ 1) 3.01(6)	Si	Ni2 ( $\times$ 1) 2.440(3)
	Ni3 (×2) 3.5551(3)		$Ce2 (\times 2) 3.8050(4)$		Ce1 $(\times 3)$ 3.21(2)		Ce1 ( $\times$ 2) 3.085(2)
	Ce1 (×4) 3.5551(7)		Ce1 ( $\times$ 1) 4.0298(7)				Ce2 (×2) 3.125(3)
	Ce2 ( $\times$ 2) 3.7074(4)		Ce1 $(\times 1)$ 4.0337(5)	Ni2	Si (×3) 2.440(3)		Ce1 $(\times 1)$ 3.160(3)
	$Ce2 (\times 1) 4.0298(7)$		Ce2 (×2) 4.303		Ce2 ( $\times$ 6) 3.0204(2)		Ce2 ( $\times$ 2) 3.169(3)
	Ce2 (×1) 4.0337(5)						Ce2 ( $\times$ 1) 3.297(2)
	Ce1 (×2) 4.303	Ni1	$[Ni3 (\times 1) 0.65(4)]$	Ni3	$[Ni1 (\times 2) 0.65(4)]$		
			[Ni1 (×1) 0.86(6)		$[Ni1 (\times 2) 1.51(4)]$		
Ce2	Ni2 (×2) 3.0204(2)		[Ni1 (×1) 1.29(6)]		[Ni3 (×2)2.152]		
Ce <sub>6</sub> Co <sub>1.</sub>	67Si3						
Cel	$[Co4 (\times 1) 2.08(2)]$		Si (×2) 3.152(5)		Ce1 ( $\times$ 3) 3.12(4)		[Co1 (×2) 1.06(6)
	$Co3 (\times 1) 2.8098(8)$		Si $(\times 1)$ 3.299(4)		$Co3(\times 1) 3.50(9)$		$[Co4 (\times 2) 1.27(3)]$
	$Co1 (\times 2) 2.91(2)$		$Ce2(\times 2) 3.6411(12)$		$Co4(\times 3) 3.57(8)$		$[Co1 (\times 2) 1.55(8)]$
	Si (×2) 3.079(3)		Ce1 $(\times 2)$ 3.7038(10)		$Co1(\times 1) 3.67(12)$		$[Ce1 (\times 1) 2.08(2)]$
	Col (×2) 3.12(4)		$Ce2(\times 2)$ 3.8178(11)		Ce1 $(\times 3)$ 4.04(6)		$[Co3 (\times 2) 2.254(10)]$
	Si (×1) 3.143(5)		Ce1 $(\times 1)$ 4.042(2)				$[Co4 (\times 4) 2.254(15)]$
	Co4 (×1) 3.22(2)		Ce1 $(\times 1)$ 4.0590(13)	Co2	Si (×3) 2.449(6)		Co4 (×2) 2.588(14)
	Co4 (×1) 3.26(2)		$Ce2 (\times 2) 4.2624$		$Ce2 (\times 6) 2.9935(4)$		Co1 (×2) 2.99(8)
	$Co4 (\times 2) 3.29(2)$						Cel $(\times 1)$ 3.22(2)
	$Co4 (\times 2) 3.32(2)$	Col	[Co1 (×1) 0.60(12)]	Co3	$[Co4 (\times 3) 0.73(3)]$		Cel ( $\times$ 1) 3.26(2)
	$Co3 (\times 2) 3.5266(6)$	001	$[Co3 (\times 1) 0.77(9)]$	000	$[Co1 (\times 2) 0.77(9)]$		Ce1 ( $\times$ 2) 3.29(2)
	Cel ( $\times$ 4) 3.527(2)		$[Co4 (\times 3) 1.06(6)]$		$[Co1 (\times 2) 1.36(9)]$		Cel ( $\times$ 2) 3.32(2) Cel ( $\times$ 2) 3.32(2)
	$Ce2 (\times 2) 3.7038(10)$		$[Co3 (\times 1) 1.36(9)]$		$[Co3 (\times 2) 2.1312]$		$Co1 (\times 2) 3.57(8)$
	$Co1 (\times 2) 4.04(6)$		$[Co1 (\times 1) 1.53(12)]$		$[Co4 (\times 6) 2.254(10)]$		
	$Ce2 (\times 1) 4.042(2)$		$[Co4 (\times 3) 1.55(8)]$		Cel ( $\times$ 3) 2.8098(8)	Si	Co2 (×1) 2.449(6)
	$Ce2 (\times 1) 4.0590(13)$		$[Co1 (\times 2) 2.13(12)]$		$Co1 (\times 2) 2.90(9)$	51	Cel ( $\times$ 2) 3.079(3)
	Cel ( $\times$ 2) 4.2624		$Col (\times 1) 2.73(12)$		$Co1 (\times 2) 3.50(9)$		$Ce2 (\times 2) 3.112(5)$
	CCI ( × 2) 4.2024		$Co3 (\times 1) 2.90(9)$		Cel ( $\times$ 6) 3.5266(6)		Cel ( $\times$ 1) 3.143(5)
Ce2	Co2 (×2) 2.9935(4)		Ce1 ( $\times$ 3) 2.91(2)		cer (× 0) 5.5200(0)		Ce2 ( $\times$ 2) 3.152(5)
CC2	$Si(\times 2) 3.112(5)$		$Co4 (\times 3) 2.99(8)$	Co4	[Co3 (×1) 0.73 (3)		$Ce2 (\times 2) 3.132(3)$ $Ce2 (\times 1) 3.299(4)$
C N:							
<i>Ce</i> <sub>5</sub> <i>Ni</i> <sub>1.8</sub> Ce1	Ni3 ( $\times$ 1) 2.8217(5)		Ce4 $(\times 1)$ 3.8450(3)		Ce2 (×3) 3.8450(3)		Ni2 (×2) 3.2026
Cer	Ni2 ( $\times$ 2) 3.0169(5)		Ce4 $(\times 1)$ 5.8450(5) Ce1 $(\times 1)$ 4.0577(5)		Ce2 ( $\times$ 3) 3.8430(3) Ce3 ( $\times$ 3) 3.9000(2)		Cel ( $\times$ 6) 3.5384(6)
	Si2 ( $\times$ 2) 3.068(2)		Ce3 ( $\times$ 1) 4.0868(3)		Ce4 ( $\times$ 2) 4.2701		CCI(x 0) 5.5504(0)
	$Si2 (\times 2) 3.008(2)$ Si2 (×1) 3.1289(13)		Ce2 ( $\times$ 2) 4.2701		CC4 ( × 2) 4.2701	Si1	Ni1 (×1) 2.3782(14)
	Cel ( $\times$ 4) 3.5384(6)		CC2 (× 2) 4.2701	Nil	Si1 (×1) 2.3782(14)	511	Ni1 ( $\times$ 1) 2.3782(14) Ni1 ( $\times$ 1) 2.393(2)
	Ni3 ( $\times$ 2) 3.5384(0)	Ce3	Ni1 (×2) 3.0099(8)	1111	$Si1 (\times 1) 2.3732(14)$ Si1 (× 1) 2.393(2)		$Ce2 (\times 2) 3.0849(9)$
	Ce2 ( $\times$ 2) 3.7039(4)	CC5	Si1 ( $\times$ 2) 3.0998(13)		$Si1 (\times 1) 2.393(2)$ Si2 (×1) 2.448(2)		Ce3 ( $\times$ 2) 3.0998(13)
			. , . ,		$Ce2 (\times 2) 2.9711(8)$		. , . ,
	Ce2 ( $\times$ 1) 4.0577(5) Ce3 ( $\times$ 1) 4.0652(7)		Si2 ( $\times$ 2) 3.131(2) Si1 ( $\times$ 1) 3.311(2)		Ce2 ( $\times 2$ ) 2.9/11(8) Ce3 ( $\times 2$ ) 3.0099(8)		Ce4 ( $\times$ 2) 3.197(2) Ce3 ( $\times$ 1) 3.311(2)
			$Ce2 (\times 1) 3.6547(5)$		Ce4 ( $\times$ 2) 3.2062(9)		Ces(x 1) 5.511(2)
	Ni2 ( $\times$ 2) 4.2683(3)				Ce4(x 2) 3.2002(9)	6:0	NE1 ( $>1$ ) 2 449(2)
	Ce1 (×2) 4.2701		Ce3 ( $\times$ 2) 3.7934(5)	NEO	[N]:2(2) + 0(75]	Si2	Ni1 ( $\times$ 1) 2.448(2)
Ce2	Ni1 $(\times 2)$ 2.0711(8)		Ce2 ( $\times$ 2)3.8217(5) Ce4 ( $\times$ 1) 3.9000(2)	Ni2	[Ni3 ( $\times$ 2) 1.0675] [Ni2 ( $\times$ 2) 2.1351]		Cel ( $\times$ 2) 3.068(2) Cel ( $\times$ 1) 3.1289(13)
	Ni1 ( $\times$ 2) 2.9711(8) Si1 ( $\times$ 2) 2.0840(0)				$[Ni2 (\times 2) 2.1351]$		Cel $(\times 1)$ 3.1289(13)
	Si1 ( $\times$ 2) 3.0849(9) Si2 ( $\times$ 2) 2.1705(15)		Ce1 ( $\times$ 1) 4.0652(7)		Cel ( $\times$ 6) 3.0169(5)		Ce3 ( $\times$ 2) 3.131(2) C=2 ( $\times$ 2) 2.1705(15)
	Si2 ( $\times$ 2) 3.1705(15) Si2 ( $\times$ 1) 2.204(2)		Ce2 ( $\times$ 1) 4.0868(3)		Ni3 (×2) 3.2026		Ce2 ( $\times$ 2) 3.1705(15)
	Si2 ( $\times$ 1) 3.294(2)		Ce3 (×2) 4.2701	NI:2	$D_{12}$ ( $\sim 2$ ) 1.0(75)		Ce2 ( $\times$ 1) 3.294(2)
	Ce3 ( $\times$ 1) 3.6547(5) Ce1 ( $\times$ 2) 3.7030(4)	$C_{2}A$	$S(1)(x,6) \ge 107(2)$	Ni3	$[Ni2 (\times 2) 1.0675]$		
	Ce1 ( $\times$ 2) 3.7039(4)	Ce4	Si1 ( $\times$ 6) 3.197(2)		Ni3 ( $\times$ 2) 2.1351		
	Ce3 ( $\times$ 2) 3.8217(5)		Ni1 ( $\times$ 6) 3.2062(9)		Cel (×3) 2.8217(5)		

Distances listed in brackets involve partially occupied sites and do not need to occur.

a new position appears in the electron density around the Co3 position. All the refined Co positions are clearly observed on the Fourier map (Fig. 7 (right)). One can notice that the electron density is less spread over the (00z)-axis and is more localised around the Co positions.

The structure of  $Ce_6Co_{1.67}Si_3$  is close to that of  $Ce_6Ni_{1.67}Si_3$  with two main differences concerning the electron density along the (00*z*)-axis. The first difference is the existence of a new Co4 position and the second one is the more important localisation of the electron density

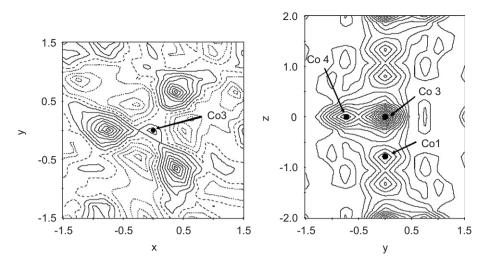


Fig. 7. (left) xy section at z = 1/4 of the difference-Fourier map for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> before the introduction of Co4 position (contour lines are in intervals of 1 (full lines) and  $-1 e \text{ Å}^{-3}$  (dashed lines)) and (right) yz section at x = 0 of the Fourier map for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> after final refinement (contour lines are in intervals of 2 (full lines) and  $-2 e \text{ Å}^{-3}$  (dashed lines)); the filled circles correspond to the refined Co positions.

around the cobalt positions. The distance between Co4 and Ce1 is too short (2.08 Å) and as previously discussed to manage realistic Co–Ce distances, a displacement of Ce1 is expected. This displacement is reflected in the high values of  $U^{11}$  and  $U^{22}$  of Ce1 ADP's matrix (Table 3). The delocalisation is stronger with Co atoms, then the  $U^{ij}$  components for Ce1 in the *ab* plane are higher to the ones observed for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>.

It is interesting to note that in the sequence  $Ce_6Ni_{1.67}Si_3 \rightarrow Ce_6Co_{1.67}Si_3$ , the *a* parameter decreases weakly (-0.1%) whereas the *c* parameter decreases more strongly (-0.9%) (Table 1). Similar remark was made for the sequence  $Ce_2NiSi_3 \rightarrow Ce_2CoSi_3$ ; the crystal structure of these ternary silicides is described also by the stacking along the *c*-axis of [Ce<sub>6</sub>] trigonal prism where Ni or Co and Si atoms are located [11–13]. In other words, in these two families of ternary silicides, the [Ce<sub>6</sub>] trigonal prisms containing Co atom are much compressed than those surrounding Ni atom; this is connected to the higher decrease of the *c* parameter.

# 3.2.3. Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub>

The extinction conditions observed for Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> agree with the  $P6_3/m$  space group already used for the refinement of the homologous compounds La<sub>5</sub>Ni<sub>1.75</sub>Si<sub>3</sub> and Pr<sub>5</sub>Ni<sub>1.9</sub>Si<sub>3</sub> [2,8]. The starting atomic positions were those reported for La<sub>5</sub>Ni<sub>1.75</sub>Si<sub>3</sub>. The same ordered distribution between Ni and Si atoms has been observed on the 6*h* sites (Table 2). The disordered positions of Ni along the chain (00*z*) are not exactly identical. The Ni3 position is still in (0 0 1/4) but Ni2 position is now on the 2*b* site (000). The electron density along this chain is still strongly delocalised and this is reflected by the very high value of  $U^{33}$  of the ADP matrix for Ni2 and Ni3 positions (Table 2), 0.117 Å<sup>2</sup> (the *U* matrixes were constraints to be the same because of high correlations).

As previously observed for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>, the electron density along the (00z)-axis is strongly delocalised and a small delocalisation of the Ni3 position in the (a,b) plane has been also observed. The strong correlations in the refinement between ADP matrices and occupancy factors did not allow us to use a non-harmonic development of the ADP tensors to describe this last delocalisation. Two maxima are clearly observed in the Fourier map (calculated at the end of the refinement) displayed in Fig. 8 (left) and correspond to the Ni2 and Ni3 positions. These maxima are also observed in the difference-Fourier map when the structure is refined without Ni atoms along the (00z)-axis (Fig. 8 (middle)). The j.p.d.f. map calculated at the end of the refinement for the Ni2 and Ni3 positions confirms that our model is in very good agreement with this observation (Fig. 8 (right)). A refinement with Ni2 in position (00z)(with z = 0.078(6)) led to similar reliability factors and electron residues but with different j.p.d.f. maps for Ni atoms along the (00z) axis; no maximum in the j.p.d.f. map is observed in the (000) position. This induces that with the model used for La<sub>5</sub>Ni<sub>1.75</sub>Si<sub>3</sub> and Pr<sub>5</sub>Ni<sub>1.9</sub>Si<sub>3</sub> by previous workers, the calculated distances with Ni2 position in (00z)would have no real significance. As our and their structural models are different, it is difficult to compare the occupancy factors refined for the Ni2 and Ni3 positions. In Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> more Ni atoms are located on Ni3 than on Ni2 position. If this is also the case for La<sub>5</sub>Ni<sub>1.75</sub>Si<sub>3</sub>, the reverse situation is observed for Pr<sub>5</sub>Ni<sub>1.9</sub>Si<sub>3</sub>.

If one considers an ideal distribution of Ni atoms only on Ni3 positions, a similar model to the one proposed by Lemaire et al. for  $Ho_6Co_{1.61}Co_3$  can be imagined: Two consecutive positions are filled by Ni atoms with a little shift away from the ideal position to increase the Ni3–Ni3 distance (2.1351 Å) and these kind of pairs of atoms would be separated by one empty position. This simple model would lead to the ideal composition  $Ce_5Ni_{1.833}Si_3$  that is

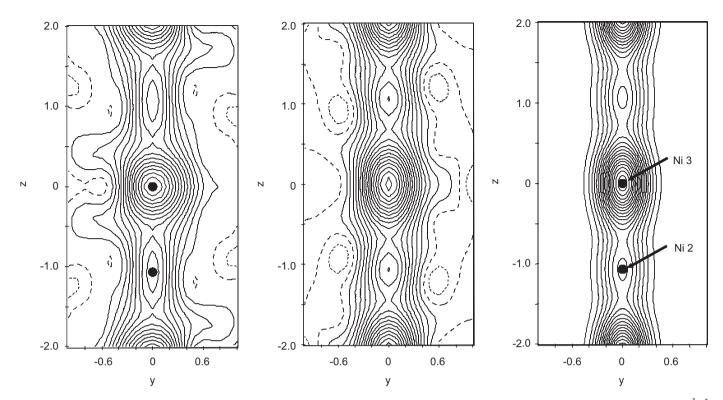


Fig. 8. Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub>: *yz* sections at x = 0 of the (left) Fourier map after final refinement (contour lines are in intervals of 2.5 (full lines) and  $-2.5 \text{ e} \text{ Å}^{-3}$  (dashed lines)), (middle) difference-Fourier map before introduction of Ni2 and Ni3 positions (contour lines are in intervals of 1 (full lines) and  $-1 \text{ e} \text{ Å}^{-3}$  (dashed lines)) and (right) joint probability maps of Ni after final refinement (contour lines are in intervals of 0.14 Å<sup>-1</sup>). The filled circles correspond to the refined Ni positions.

very close to the refined stoichiometry. Moreover, the necessity to increase the Ni3–Ni3 distances would explain the strong deformation of the electron density around the Ni3 position.

A projection of the crystal structure of Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> is displayed in Fig. 6. This structure is the member n = 3 of the series  $RE_{(n+1)(n+2)}Ni_{n(n-1)+2}Si_{n(n+1)}$  [2]. Nine chains of face-sharing trigonal prism of cerium [Ce<sub>6</sub>] form a triangle. These chains filled by Ni1 and Si atoms are connected together through shared faces. The trigonal prisms surrounding Si1 and Si2 atoms are mono- and bi-capped by Ce atoms, respectively. Nil-Ce average distance is equal to 3.062 Å and Si-Ce distances range from 3.085 to 3.311 Å (Table 4). All these distances are in good agreement with the sum of the metallic radii, 3.071 and 3.144 Å for  $r_{\rm Ni} + r_{\rm Ce}$ and  $r_{\rm Si} + r_{\rm Ce}$ , respectively. Between the triangles of chains of [Ce<sub>6</sub>] trigonal prisms, a chain of face-shared octahedra of Ce is filled by Ni2 and Ni3 atoms. Ni2 atoms are exactly at the centre of the octahedra with Ni2-Ce1 distance of 3.0169 Å. These distances are relatively short and this would be correlated to the low occupancy of this 2bsite and the strong delocalisation of the Ni atoms in this chain. Ni3 positions are on the face of the octahedra in a tri-capped elongated trigonal prism with six long Ni3-Cel distances (3.5384Å) and three short Ni3-Cel distances (2.8217 Å). In this crystal structure, Cel atoms exhibit also high ADP in ab plane to manage the steric strains (Table 3).

## 3.3. Magnetic properties

The temperature dependence of the reciprocal magnetic susceptibility  $\chi_m^{-1}$  of  $Ce_6Ni_{1.67}Si_3$  and  $Ce_6Co_{1.67}Si_3$ , measured with an applied field of 4T, is displayed in Fig. 9. Above 150 K, these  $\chi_m^{-1} = f(T)$  curves follow a Curie–Weiss law. The experimental value of the effective magnetic moment  $\mu_{eff.}$  is 2.45  $\mu_B/Ce$  for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> and 2.59  $\mu_B/Ce$ for  $Ce_6Co_{1.67}Si_3$ , which is close to the calculated one for a free Ce<sup>3+</sup> ion (2.54  $\mu_{\rm B}$ /Ce). At low temperature than 150 K, the  $\chi_m^{-1} = f(T)$  curves deviate from a straight-line behaviour, presumably due to thermal depopulation of crystalfield levels. The paramagnetic Curie temperature  $\theta_p$  is found to be -36 K for  $\text{Ce}_6 \text{Ni}_{1.67} \text{Si}_3$  and -77 K for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>. Such a large negative value of the paramagnetic Curie temperature is often found in Kondo compounds. It is important to note that  $\theta_p$  is higher for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> than for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>. Similar increase of  $\theta_p$ was observed in the sequence  $Ce_2CoSi_3$  ( $\theta_p = -107 \text{ K}$ )  $\rightarrow$ Ce<sub>2</sub>NiSi<sub>3</sub> ( $\theta_p = -22$  K) [13]; the structural properties of these last ternary compounds show, as mentioned above, many similarities with Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> and Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>.

No magnetic ordering can be detected above 1.8 K by magnetisation measurements performed on  $\text{Ce}_6\text{Co}_{1.67}\text{Si}_3$  (Fig. 10(a)); its magnetisation becomes monotonically decreasing function of the temperature without any indication of the ordering. On the contrary, different behaviour is observed for  $\text{Ce}_6\text{Ni}_{1.67}\text{Si}_3$  (Fig. 10(b)); its

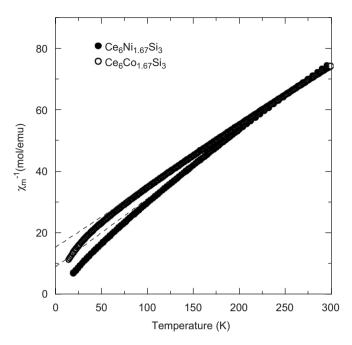


Fig. 9. Temperature dependence of the reciprocal magnetic susceptibility of  $Ce_6Ni_{1.67}Si_3$  and  $Ce_6Co_{1.67}Si_3$ . The dashed lines present the Curie–Weiss law.

magnetisation (i) is higher than that observed for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> and (ii) tends to saturate below  $T_t = 3.8(2)$  K (the temperature where the derivative curve  $d(M/\mu_0H)/dT = f(T)$  exhibits a maximum). This preliminary study suggests the establishment of long-range magnetic correlations below  $T_t = 3.8(2)$  K in Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>. More information obtained for instance by ac-magnetic susceptibility measurements or neutron powder diffraction are necessary in order to understand the nature of the magnetic ordering appearing below  $T_t$  in this last ternary silicide.

These different magnetic behaviours between Ce<sub>6</sub>Co<sub>1.67</sub>  $Si_3$  and  $Ce_6Ni_{1.67}Si_3$  are clearly evidenced by specific heat measurements (Fig. 11). No anomaly can be distinguished from the  $C_p = f(T)$  curve concerning Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> in agreement with the absence of magnetic ordering above 1.8 K as observed by magnetisation measurements. But below 6 K, the curve  $C_p/T = f(T)$  (inset of Fig. 11) exhibits an upturn attaining a value of  $389 \text{ mJ/Ce-mol } \text{K}^2$  at 1.9 Kas if the compound is a heavy fermion; alternatively, a possible magnetic ordering well below 1.9 K also can contribute to the low-temperature enhancement of  $C_p/T$ . It may be mentioned that between 10 and 22 K, we can obtain the electronic coefficient  $\gamma_{LT}$  by fitting the data to the formula  $C_{\rm p}/T = \gamma_{\rm LT} + \beta_{\rm LT}T^2$ . The obtained values were  $\gamma_{LT} = 162 \text{ mJ/Ce-mol } \text{K}^2 \text{ and } \beta_{LT} = 3.47 \, 10^{-4} \, \text{J/Ce-mol } \text{K}^4.$ Considering this value of  $\gamma_{LT}$ , we could define this ternary silicide as a moderate heavy fermion system. Finally, we note that this  $\gamma_{LT}$  value is higher than that reported for the intermediate valence compounds  $Ce_2CoSi_3 (\gamma_{LT} = 59.7 \text{ mJ}/$ Ce-mol K<sup>2</sup>) [13,15,16]. On the contrary, the curve  $C_p = f(T)$ for Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> (Fig. 11) exhibits a peak at about 4.1(2) K and a peak value approaching 5.36 J/Ce-mol K. The temperature of the heat-capacity peak is practically the

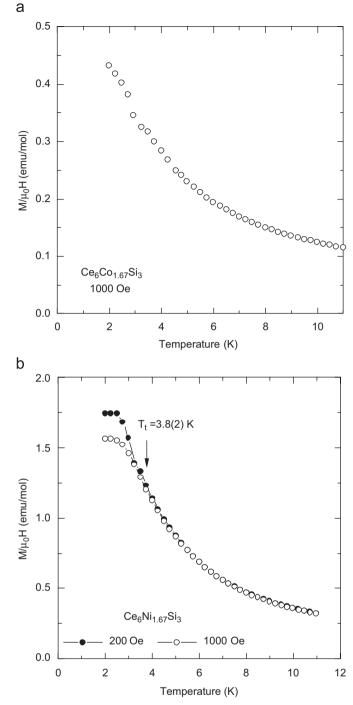
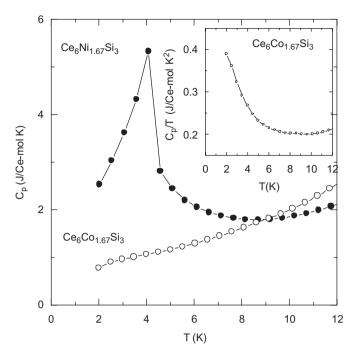


Fig. 10. Temperature dependence of the magnetisation divided by applied field for: (a)  $Ce_6Co_{1.67}Si_3$  ( $\mu_0H = 1000 \text{ Oe}$ ) and (b)  $Ce_6Ni_{1.67}Si_3$  ( $\mu_0H = 200$  and 1000 Oe).

same as the temperature where the derivative curve of the magnetisation  $d(M/\mu_0 H)/dT = f(T)$  exhibits a maximum (Fig. 10(b)) confirming the occurrence of magnetic ordering below  $T_t$ . Valuable information can be obtained from the entropy associated with the magnetic ordering of Ce<sub>6</sub>Ni<sub>1.67</sub> Si<sub>3</sub>, which is estimated from the magnetic contribution to the specific heat in the low-temperature range as  $C_{\rm pmag} = C_{\rm p} - (\gamma T + \beta T^3)$  (between 10 and 22 K, the fitting of



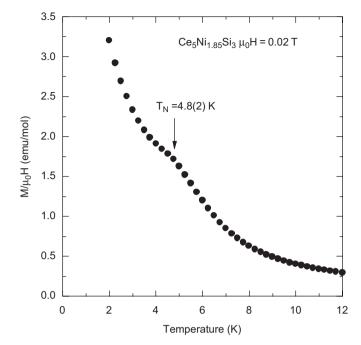


Fig. 11. Temperature dependence of the specific heat  $C_p$  of Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> and Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>. The curve  $C_p/T = f(T)$  for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> is plotted in the inset.

Fig. 12. Temperature dependence of the magnetisation divided by applied field for  $Ce_5Ni_{1.85}Si_3$  ( $\mu_0H = 200 \text{ Oe}$ ).

 $C_{\rm p}/T = \gamma + \beta T^2$  yields an electronic coefficient  $\gamma = 144 \text{ mJ}/\text{Ce-mol K}^2$  smaller than that determined for Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub> and phonon constant  $\beta = 2.29 \, 10^{-4} \text{ J/Ce-mol K}^4$ ). At  $T_{\rm t}$ , the magnetic entropy reaches  $3.75 \, \text{J/Ce-mol K} = 0.65 \, \text{RLn2}$  which is smaller than RLn2 =  $5.76 \, \text{J/Ce-mol K}$ , the value of magnetic entropy expected for a doublet ground state of Ce<sup>3+</sup>. This reduction suggests the presence of a moderate Kondo effect in this ternary silicide Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub>.

Above 150 K, the curve  $\chi_m^{-1} = f(T)$  (not shown here) for Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> follows a Curie–Weiss law with  $\mu_{eff} = 2.51 \,\mu_{B}/$ Ce and  $\theta_p = -43$  K as effective moment and paramagnetic Curie temperature. These values are comparable to those reported for the ternary silicide Ce<sub>5</sub>Ni<sub>2</sub>Si<sub>3</sub>:  $\mu_{eff} = 2.42 \,\mu_{B}/$ Ce and  $\theta_p = -61.3 \text{ K}$  [14]. Considering both the previous works devoted to La<sub>5</sub>Ni<sub>1.75</sub>Si<sub>3</sub> [2] and Pr<sub>5</sub>Ni<sub>1.9</sub>Si<sub>3</sub> [6] and the present study on  $Ce_5Ni_{1.85}Si_3$ , we assume that the  $Ce_5Ni_2Si_3$  alloy corresponds to the single phase  $Ce_5Ni_{1.85}$ Si<sub>3</sub>. We note that the unit cell parameters determined here for this last ternary silicide (Table 1) are close to those reported for Ce<sub>5</sub>Ni<sub>2</sub>Si<sub>3</sub> (a = 15.997(9) and c = 4.309(1)Å) [11,14]. At low temperature, the thermal dependence of the magnetisation divided by applied field of Ce<sub>5</sub>Ni<sub>1.85</sub>Si<sub>3</sub> shows a kink at  $T_{\rm N} = 4.8(2)$  K (Fig. 12). This behaviour suggests that this ternary silicide is antiferromagnetically ordered. This  $T_N$ -temperature is smaller than that reported previously for Ce<sub>5</sub>Ni<sub>2</sub>Si<sub>3</sub> [14].

# 4. Conclusion

This study reveals that the ternary silicides  $Ce_6Ni_2Si_3$ and  $Ce_5Ni_2Si_3$  reported previously [1,11,14] cannot be obtained as single phase. On the contrary, the microprobe analysis of these alloys allows the characterisation of two new ternary compounds Ce<sub>6</sub>Ni<sub>167</sub>Si<sub>3</sub> and Ce<sub>5</sub>Ni<sub>185</sub>Si<sub>3</sub>. Their investigation by X-ray diffraction on single crystal shows a strong delocalisation of the electron density of the Ni in the infinite chain of face-shared octahedra of cerium. Similar crystallographic consideration was evidenced on the new compound based on cobalt Ce<sub>6</sub>Co<sub>1.67</sub>Si<sub>3</sub>. In this last compound, a higher delocalisation of the Co atoms in the [Ce1] triangle has been observed and leads to the existence of an extra position (Co4 position) very close to the cerium atoms (Cel position). This induces a strong deformation of the thermal ellipsoid of the Ce1 position. The preliminary study of their physical properties indicates that Ce<sub>6</sub>Ni<sub>1.67</sub>Si<sub>3</sub> orders magnetically below  $T_t = 3.8(2)$  K whereas Ce<sub>6</sub>Co<sub>1.67</sub> Si<sub>3</sub> exhibits a heavy fermion behaviour. More information obtained for instance by electrical resistivity measurement and neutron powder diffraction are necessary in order to establish a relationship between the atomic disorder existing in these phases and their physical properties.

## Acknowledgments

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#### Appendix A. Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.jssc.2007.01.030.

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